

# Chapter 11 Review Gases Section 2

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Chapter 11 Review Gases Section SECTION 1 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c ripld. 7-0 (d) unchanged. Rank the following pressures in increasing order. (c) 76 torr (a) 50 kPa O, OOìctbv-x Home - Kenilworth Public Schools CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase a. temperature increases, volume stays the same decrease b. volume increases, temperature stays the same mc06se cFMsr i-vi - clarkchargers.org Chapter 11 Review Gases Section SECTION 1 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c ripld. 7-0 (d) unchanged. Rank the following pressures in increasing order. Chapter 11 Review Gases Section 4 Answers Chapter 11 Gases. Chapter11. Gases. 459. t's Monday morning, and Lilia is walking out of the chemistry building, thinking about the introductory lecture on gases that her instructor just presented. Dr. Chapter 11 Gases - An Introduction to Chemistry Chapter 11 Test Review. multiple choice (25) definition & applications of pressure (also atmospheric) SI unit of force. definition

& use of a barometer. standard temperature & pressure (STP) Definition of Dalton's law of partial pressures. Definitions & formulas for Boyle's, Charles', Gay-Lussac's, and combined gas laws Modern Chemistry Chapter 11 GASES Chapter 11 - Gases - An Introduction to Chemistry 182 Study Guide for An Introduction to Chemistry Section Goals and Introductions ... Section 11.3 Equation Stoichiometry and Ideal Gases ... Chemistry Section 11 Answers - Free PDF File

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multiplied by 22.4. L. CHAPTER REVIEW Gases Hydrogen molecule 1 mol H<sub>2</sub> at STP = 22.4 L Oxygen molecule 1 mol O<sub>2</sub> at STP = 22.4 L Carbon dioxide molecule 1 mol CO<sub>2</sub> at STP = 22.4 L CHAPTER 11 Molecular Composition of Gases Start studying Chapter 11 Molecular Composition of Gases. Learn vocabulary, terms, and more with flashcards, games, and other study tools. Chapter 11 Review Molecular Composition Of Gases Mixed Answers Related with Chapter 11 Review ' To Kill A Mockingbird - Edmond Public Schools (5,207 View) Chapter 1 Chapter 2 Chapter 3 Chapter 4 Chapter 5 (1,384 View) Chapter Chapter Test A 6 For Use After Chapter 6 (988 View) Review Unit: Chemistry Review - Nelson Education (1,850 View) Chapter 3 Test Review - Goochland County Public Schools (3,007 View)

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